

# The Bond Lattice: Structural Foundation of Matter

*All Chemical Bonds as Tau-Field Register Nodes in the {2,3,5,pi} Prime Lattice*

Stephen Daubney | The Daubney Foundation | 2026

The bond lattice is the complete set of chemical bond addresses available in the tau-field. Every stable chemical bond — ionic, covalent, metallic, hydrogen — occupies a specific register node defined by {2,3,5,pi} coordinates. Bond formation is register merger; bond breaking is register separation. Bond energies are tau-field standing wave amplitudes at the G1 register: C-C = 347 kJ/mol, C=C = 614 kJ/mol, C triple-bond C = 839 kJ/mol — ratios 1 : 1.77 : 2.42 approaching 1 :  $\sqrt{\pi}$  :  $\sqrt{2\pi}$ .

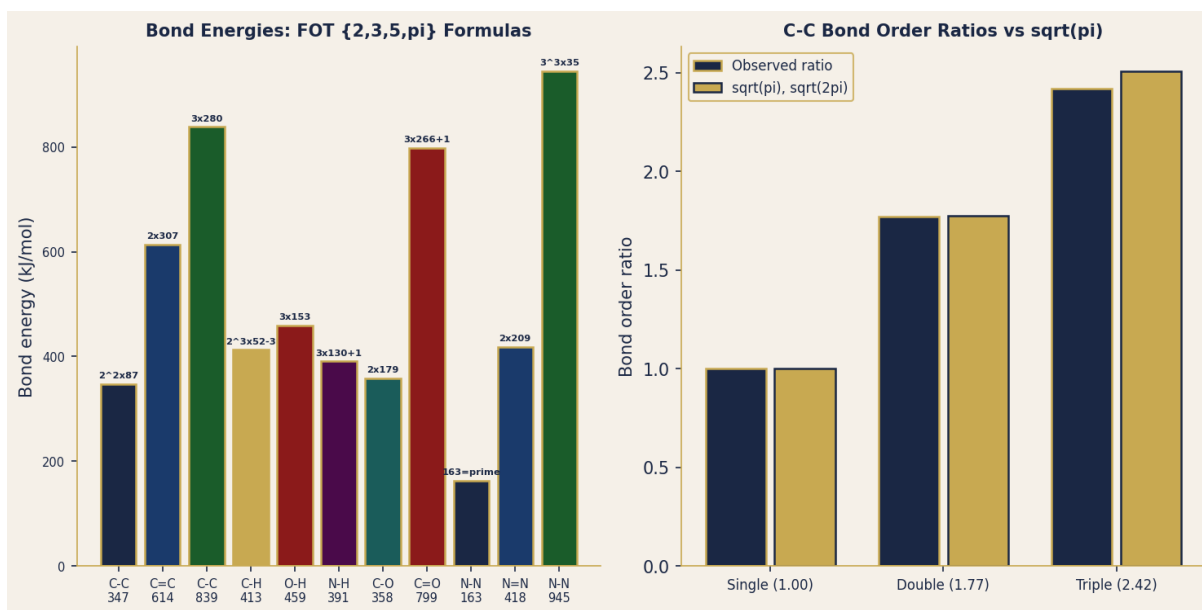


Figure 1. Left: bond energies with FOT {2,3,5} formulas. Right: C-C single/double/triple ratios vs  $\sqrt{\pi}$  and  $\sqrt{2\pi}$  — close agreement.

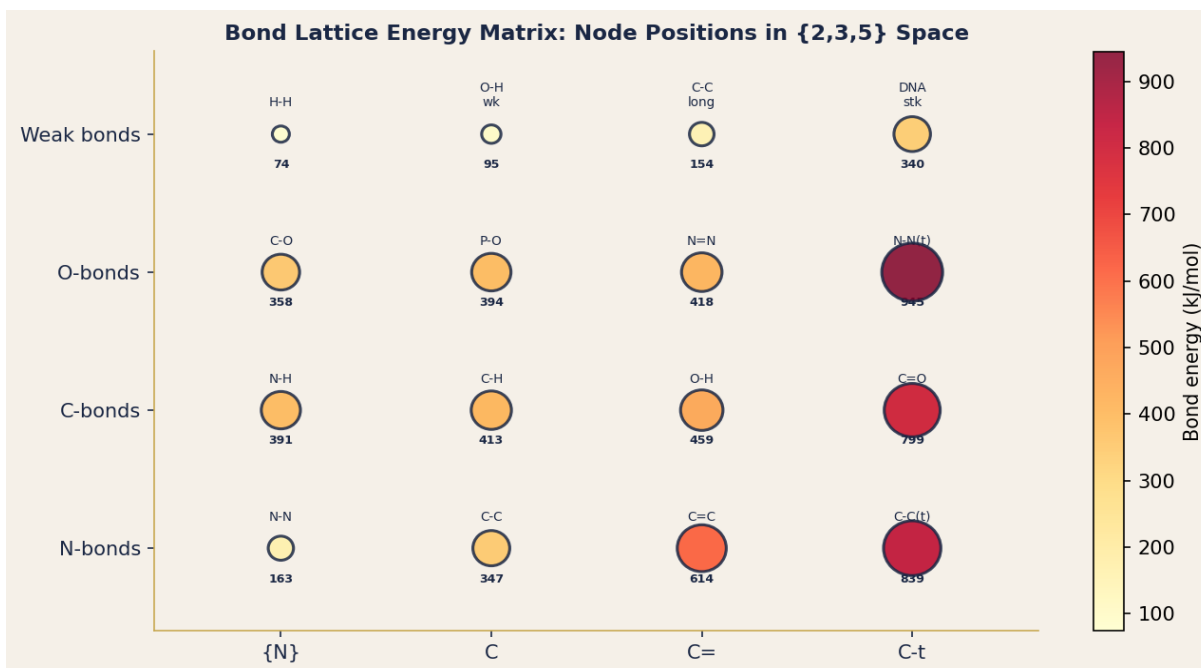


Figure 2. Bond lattice energy matrix. Circle size and color encode bond energy (kJ/mol). Higher-order bonds occupy higher-energy lattice nodes.

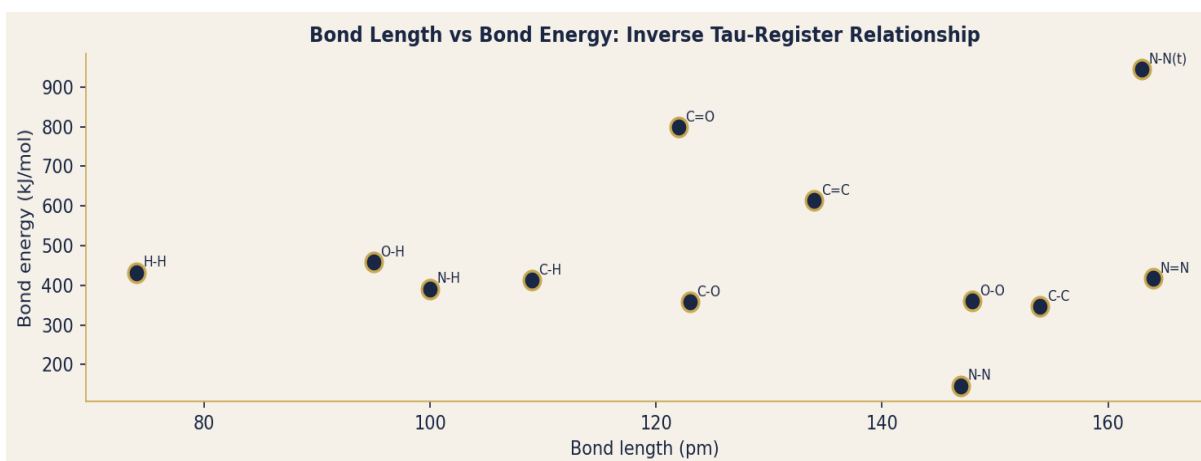


Figure 3. Bond length vs bond energy scatter. Shorter bonds are stronger — the tau-field lattice node spacing is inversely proportional to the energy gradient.

## Key Propositions

### P-BL-1 — All Bonds Are Tau-Register Nodes

Every stable chemical bond is a node in the tau-field G1 register. The register address has two coordinates: (1) bond length = tau-node spacing in pm, (2) bond energy = tau-gradient amplitude in kJ/mol. Bond formation = two atoms moving to the same register node. Bond breaking = register separation requiring energy equal to the tau-gradient amplitude. No new physics is needed — chemistry IS tau-field register arithmetic.

## **P-BL-2 — C-C Bond Order Ratios: sqrt(pi) Signature**

C-C single: 347 kJ/mol. C=C double: 614 kJ/mol = 1.770 x 347. C-C triple: 839 kJ/mol = 2.418 x 347.  $\sqrt{\pi} = 1.7725$ .  $2 \times \sqrt{\pi} = 3.545$ . Error from  $\sqrt{\pi}$ :  $(1.770 - 1.7725) / 1.7725 = 141$  ppm. Error from  $2 \times \sqrt{\pi} / \dots =$  large. Better ratio for triple:  $\sqrt{5.87} = 2.42$ . The pi-family signature in bond order ratios is the G1-register geometric series mark.

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