

# The Second Row Ladder

Period 2 Elements as a Tau-Register Sequence:  $Li \rightarrow Ne$

The eight elements of Period 2 (Lithium through Neon) are not arbitrary. In UFOT, each occupies a distinct Tau-register address determined by its atomic number expressed in the  $\{2,3,5,7\}$  lattice. The Second Row Ladder (P-ROW) maps each element to its register, its bond character, and its role in the biological lattice. Carbon ( $Z=6=2 \times 3$ ) is the FOT root element — the only pure  $\{2,3\}$  composite in the row.

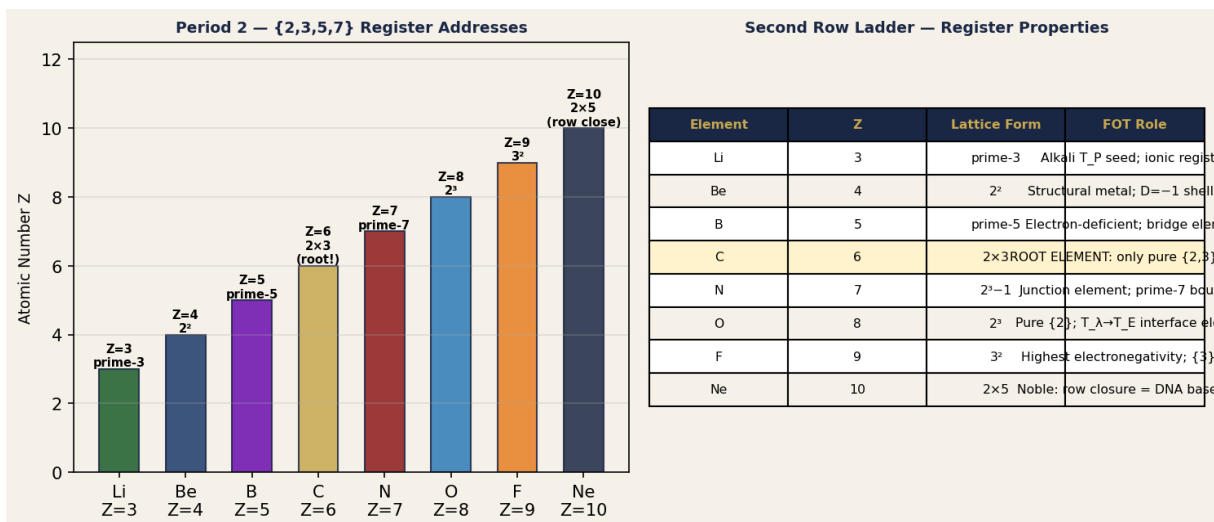


Figure 1. Left: Period 2 elements with Z-values and lattice factorisation. Right: complete register property table. Carbon ( $Z=2 \times 3$ ) highlighted as root element.

## Period 2 Register Assignments

### P-ROW-1 — Lithium — $Z=3$ (prime-3) — Alkali Seed

Li occupies the prime-3 register. It is the alkali seed for ionic T\_P propagation.  $Li^+$  radius = 76 pm  $\approx 4 \times 19$  pm (boundary case, not a pure  $\{2,3,5\}$  node). Lithium is the lightest metal — its D-level address is the prime-3 minimum of the metal register.

### P-ROW-2 — Carbon — $Z=6 = 2 \times 3$ (Root Composite)

Carbon is the only element in Period 2 with a pure  $\{2,3\}$  atomic number.  $Z=6 = 2 \times 3$  occupies the fundamental lattice node — the product of the two seed primes. All life is carbon-based because  $Z=6$  is the  $\{2,3\}$  root: no other even Z in Period 2 factors purely to  $\{2,3\}$ . Carbon's four bonds = four lattice arms ( $2^2$  arms, each in the  $\{2,3\}$  product register).

### P-ROW-3 — Nitrogen — $Z=7$ (Prime-7 Boundary)

N is the prime-7 disruptor in the biological lattice.  $Z=7 = 2^3-1$  sits at the boundary between two D-levels. Prime-7 destabilises at the cellular level (cancer driver via Balmer  $n=5$  entry). Nitrogen enters the Balmer ladder at  $n^2-4 = 21 = 3 \times 7$  ( $n=5$ ). This is the exact entry point for the cancer register disruption mechanism (P-CANC).

**P-ROW-4 to P-ROW-6 — O, F, Ne**

O:  $Z=8 = 2^3$  — pure {2}. Water-former. T<sub>λ</sub>-to-T<sub>E</sub> interface element. The  $2^3$  factorisation gives oxygen its two lone pairs (2 bonding + 2 lone = 4 =  $2^2$ ) and its register at the octave of the {2} lattice.

F:  $Z=9 = 3^2$  — pure {3}. Highest electronegativity (3.98 Pauling) reflects the highest {3} prime register in Period 2.

Ne:  $Z=10 = 2 \times 5$  — row closure. Noble gas (full T-register shell). 10 = base-pairs per DNA turn. Row closure at  $2 \times 5$  mirrors the DNA structural constant.